

Physics: Paper - 602 (Statistical Mechanics And Solid State Physics) (New Course)

> Faculty Code: 003 Subject Code: 1016032

Time :  $2\frac{1}{2}$  Hours]

[Total Marks: 70

Seat No.

- Instructions: (1) Symbols and notations have their usual meaning.
  - Total marks of the question are indicated on (2)the right side of the question.
  - (3)Attempt as many questions as instructed in the main question.

## UNIT - 1

1 (A) Objective type questions: (1 mark each) 4

- How many coordinates does the phase space have?
- (2)The electrons are regarded as Fermions. The statement is true or false?
- Which type of statistical laws are applicable to Photons? (3)
- What is the least volume occupied by the phase cell? (4)
- Answer in brief: (any one out of two) (B)

- Suppose at N.T.P. a gas has a molecular density of  $10^{19}$  molecules/cm<sup>3</sup>, if the average radius of the molecule of gas is  $10^{-8}$  cm. show that the gas molecules are distinguishable particles.
- Find the value of 50! using Striling's approximation. (2)
- Answer in detail: (any **one** out of two)

3

- Discuss Stirling's approximation. (1)
- Compare and discuss the BS statistics and FD (2)statistics.
- Write notes on following: (any one out of two)

5

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- Discuss in detail the Maxwell-Boltzmann distribution laws.
- Discuss in detail the Fermi-Dirac distribution laws. (2)

## UNIT - 2

2	(A)	Objective type questions : (1 mark each) (1) Give the names of seven crystal systems.	4
		(2) How many Bravais lattices are possible for	
		Tetragonal system? (3) Give one example of Ionic crystal and Covalent crystal	al.
		(4) Define Heat capacity.	
	(B)	Answer in brief: (any <b>one</b> out of two)	2
		(1) In a cubic system with lattice constant 'a' calculate	
		the spacing between planes of $(1 \ 0 \ 0)$ , $(1 \ 1 \ 0)$ , $(1 \ 1 \ 1)$	
		and $(101)$	
		(2) $NaCl$ crystallizes as cubic structure. The molecular weight of $NaCl$ is $58.46 \ gm$ and the density at	
		room temperature is $2.167  gm/cm^3$ , calculate the	
		lattice constant of the cubic unit cell. (Avogadro	
		number is $6.02 \times 10^{23} mole^{-1}$ )	
	(C)	Answer in detail: (any one out of two)	3
		<ul><li>(1) Write a note on the covalent crystals.</li><li>(2) Write a note on Miller indices in crystalline solids.</li></ul>	
	(D)	Write a note on following: (any <b>one</b> out of two)	5
	(-)	(1) Write detailed note on classical theory of heat capacity of solids.	
		(2) Write detailed note on the Einstein's model for	
		heat capacity of solids.	
		UNIT - 3	
3	(A)	Objective type questions: (1 mark each)	4
		(1) What is the Wiedemann-Frantz ratio in metal? What does it signify?	
		(2) Define density of available electronic states $D(E)$ in Free electron model?	
		(3) Define Fermi energy level in the Free electron model of metals.	
		(4) Which type of statistical distribution law was applied by Sommerfield in his modification to the	
		Free electron model for metals.	
	(B)	Answer in brief: (any one out of two)	2
	(-)	(1) Calculate the number of the energy states available	
		for electrons in one cubic cm of box lying below	
		the energy of $1 eV$ .	
		(2) Evaluate the temperature at which there is 25%	
		probability that a state with energy 0.5 eV above the Fermi energy level would be occupied by	
		electron.	
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	(C)	<ul> <li>Answer in detail: (any one out of two)</li> <li>(1) Derive the equation for calculating the Fermi energy levels in the metal having volume V and N free electrons.</li> <li>(2) Explain the classification of Conductors and Insulators based on band diagram.</li> </ul>	3
	(D)	<ul> <li>Write note on following: (any one out of two)</li> <li>(1) Write a detailed note on Boltzmann transport equation.</li> <li>(2) Using the Sommerfield modification to the electronic part of the heat capacity of metal write the detailed note on the thermal conductivity of metals.</li> </ul>	5
		UNIT - 4	
4	(A)	<ul> <li>(1) The Fermi level lies half way between the valance and the conduction band in insulators. The statement is true or false?</li> <li>(2) P and N type semiconductors are called intrinsic</li> </ul>	4
		semi-conductors. True or False?  (3) The energy band gap in Germanium semiconductor is higher than that in Silicon semiconductor. True or False?  (4) The resistivity of semiconductors decreases with	
	( <b>D</b> )	temperature. True or False?	•
	(B)	Answer in brief: (any <b>one</b> out of two)  (1) If the Fermi level in n-type semi-conductor lies  0.4 eV below the conduction band. What is the proportion of the donor impurity compared to the total charge carried at 300K temperature.	2
		(2) For an intrinsic semiconductor with $E_g = 0.7  eV$ determine the position of the Fermi level at $300K$ if $m_p^* = 6m_e^*$ .	
	(C)	<ul> <li>Answer in detail: (any one out of two)</li> <li>(1) Explain in brief the bonding in semiconductor.</li> <li>(2) Explain effect of impurities on intrinsic semiconductor.</li> </ul>	3
	(D)	(1) Discuss carrier concentration in intrinsic semi- conductor	5
ID	000 1	(2) Discuss Fermi level in extrinsic semiconductor.	
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## UNIT - 5

5	(A)	Objective type questions: (1 mark each)	4
		(1) The superconducting state was first observed in	
		which material?	
		(2) Which type of magnetism is seen in the	
		superconducting state?	
		(3) What do the terms $T_c$ and $H_c$ denote in	
		superconductivity?	
		(4) Cupper pairing is mediated by phonons.	
		The statement is true or false?	
	(B)	Answer in brief: (any one out of two)	2
		(1) If the density of Lead is $11.3 \times 10^3 kg/m^3$ and	
		atomic weight is 207.19, then calculate the London	
		penetration depth at $0 K$ .	
		(2) For Aluminium the $H_0 = 99 Gauss$ and	
		$T_c = 1.18K$ , then calculate the value of critical	
		field $H_c$ at the temperature $T=0.5K$	
	(C)	Answer in detail: (any one out of two)	3
		(1) Explain in brief the influence of external agents	
		on superconducting state.	
		(2) Explain in brief the London theory of	
		superconductivity.	
	(D)	Write notes on following: (any one out of two)	5
		(1) Write detailed note on the thermodynamics of	
		superconductivity.	
		(2) Write detailed note on the Josephson effect in	
		superconductivity.	